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RareBooksClub. Paperback. Book Condition: New. This item is printed on demand. Paperback. 98 pages. Original publisher: Washington : U. S. G. P. O. , 2008. LC Number: KF27 . S3935 2008m OCLC Number: (OCoLC)312483883 Subject: Water acidification -- Research -- United States -- Finance. Excerpt: . . . 11 And then that changes the pH of the ocean. Is that the basic concern mechanically of what is going on NSLEE Mr. I. Given my test scores in freshman chemistry, I may not be the proper person to ask that question to, but generally speaking, yes. I actually dont know if it is conveyed through rain or just some partial pressure driving the CO to go into solution, 2 and the chemists behind me will answer that. And I should point out that when we say it becomes more acidic, it actually becomes less base, because the ocean is actually in base conditions, and it reduces the pH, a very small number. On a logarithmic scale that is about a 30 percent increase in those ions that would be, acidic. So it is definitely a corrosive, though, for biological, anything forming a calcium-base system, and that is where the real problem is. So I hope that I didnt cloud it too much. Mr. A. Mr. Chairman, if it is okay, is there somebody else KIN who wanted to go a little further Mr. I. Yes. You better defer to the real scientific - - NSLEE AMPSON Chairman L. We will have the next panel come up in just a minute. KIN Mr. A. Thank you, Mr. Inslee. AMPSON Chairman L. They will be able to get into the details of that. Dr. Bartlett. Mr. B. Thank you very much. ARTLETT Carbon dioxide, of course, is very soluble in water, and the...



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